## Investigation of *in vitro* Antioxidant and Cytotoxic Potential of Triphala Methanol Extract in Different Ratios

By

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A thesis submitted to the Department of Pharmacy in partial fulfillment of the requirements for the degree of Bachelor of Pharmacy (Hons.)

Department of Pharmacy
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**Declaration** 

It is hereby declared that

1. The thesis submitted is my own original work while completing degree at Brac

University.

2. The thesis does not contain material previously published or written by a third party,

except where this is appropriately cited through full and accurate referencing.

3. The thesis does not contain material which has been accepted, or submitted, for any

other degree or diploma at a university or other institution.

4. I have acknowledged all main sources of help.

**Student's Full Name & Signature:** 

Shifat Sehrin Khan Upoma

## Approval

The thesis titled "Investigation of in vitro Antioxidant and Cytotoxic Potential of Triphala Methanol Extract in Different Ratios" submitted by Shifat Sehrin Khan Upoma (16146036) of Spring, 2016 has been accepted as satisfactory in partial fulfillment of the requirement for the degree of Bachelor of Pharmacy (Hons.) on 27<sup>th</sup> February, 2020.

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## **Ethics Statement**

This study does not involve any animal or human trial.

**Abstract** 

Triphala is comprised of Bibhitaki Amalaki and Haritaki and its methanol extract was

chosen for the *In vitro* antioxidant and cytotoxic potential studies. Firstly, in the DPPH

test the IC<sub>50</sub> value was found to be 15.93  $\mu$ g/ mL, 12.99  $\mu$ g/ mL and 10.287  $\mu$ g/ mL for

Haritaki, Bibhitaki and Amlaki, and the IC<sub>50</sub> value for Ascorbic Acid was 12.96 μg/ mL,

so Amlaki showed more inhibition. The IC<sub>50</sub> value for the Triphala ratios 1:1:1 (A:B:H),

2:1:1 (A:B:H), 1:2:1 (A:B:H) and 1:1:2 (A:B:H) were 11.28 μg/ mL, 7.01 μg/ mL, 8.33

μg/ mL and 11.03 μg/ mL respectively and all the values were lower than that of the

standard. Secondly, at 1200 µg/ mL, the total phenolic content was found to be 368.94 mg,

395.09 mg and 260.65 mg for Bibhitaki, Amlaki and Haritaki respectively. Finally, at 1200

μg/ mL, the total antioxidant capacity was found to be 245.32mg, 485.5mg and 559.76mg

for Bibhitaki, Amlaki and Haritaki respectively. In N4X4 cell line, IC50 value was

13.43mg/ mL and 13.23mg/ mL at 25 mg/ mL for 1:1:1(A:B:H) and 2:1:1 (A:B:H)

Triphala ratio. To conclude, it can be inferred that the extracts showed strong antioxidant

and moderate cytotoxic potential.

Keywords: Antioxidant; DPPH; Cytotoxicity; N4X4; Triphala

v

Dedication			
Dedicated to my be	cloved family member	rs, for their uncondition	nal love and support.

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## **Table of Contents**

Declarationii
Approvaliii
Ethics Statementiv
Abstractv
Dedicationvi
Acknowledgementvii
List of Tablesxiv
List of Figuresxvi
List of Acronymsxviii
Chapter 1 1
Introduction1
1.1 Medicinal plants used for phytotherapy1
1.2 Overview of Bangladeshi Medicinal Plants3
1.3 Reactive Oxygen Species: Source and Side-effects4
1.4 Antioxidants5
1.4.1 Significance of Antioxidant Potential of Plants5
1.5 In vitro Antioxidant Assay7
1.6 Brain Tumour7
1.6.1 Types of Glioma

1.6.2 Anticancer drugs from Medicinal Plants	7
1.7 In vitro Cytotoxicity Screening	9
1.7.1 MTT (3 - (4,5 – dimethylthiazol-2-yl) - 2, 5 - diphenyltetrazolium bromide) assay	, 9
1.8 Selection of <i>Triphala</i> for the Study	10
1.9 Rationale of This Study	10
1.10 Aim of This Study	11
1.11 Objectives of This Study	11
1.12 Introduction of the Plant	11
1.12.1 Phyllanthus emblica or Emblica officinalis (Amlaki)	11
1.12.1.1 General Description	11
1.12.1.2 Traditional Uses	12
1.12.1.3 Constituents	12
1.12.1.4 Taxanomical Classification	12
1.12.2 Terminalia bellirica (Bibhitaki)	13
1.12.2.1 General Description	13
1.12.2.2 Traditional Uses	13
1.12.2.3 Constituents	13
1.12.2.4 Taxonomical classification	13
1.12.3 Terminalia chebula (Haritaki)	14
1.12.3.1 General Description	14
1.12.3.2 Traditional Uses	14
1 12 3 3 Constituents	1.1

1.12.3.4 Taxonomical classification	15
Chapter 2	17
Literature Review	17
2.1 Previously Studied Pharmacological Activities of Triphala	17
2.1.1 Triphala as An Antineoplastic Agent	19
2.1.2 Triphala as A Radioprotective Effect	19
2.1.3 Triphala as A Chemoprotective Agent	20
2.1.4 Triphala as A Chemopreventive Agent	20
2.1.5 Antimicrobial Activity	20
2.1.6 Antihyperglycemic Effects	21
2.1.7 Antipyretic and Analgesic Activities	21
2.1.8 Antioxidant Effects	21
Chapter 3	22
Methodology	22
3.1 Collection and Authentication	22
3.2 Extraction Procedure	22
3.2.1 Extraction	22
3.2.2 Filtration	22
3.2.3 Concentration and Drying	23
3.3 In vitro Antioxidant Assay	23
3.3.1 DPPH (1, 1-diphenyl-2-picryl hydrazyl) Free Radical Scavenging Assay	23
3.3.1.1 Reagents and Chemicals	23

3.3.1.2 Sample and Standard Solution Preparation	24
3.3.1.3 DPPH solution preparation	24
3.3.1.4 Experimental Procedure	24
3.3.2 Determination of Total Phenolic Content (TPC)	)25
3.3.2.1 Reagents and Chemicals	
3.3.2.2 Reagent Preparation	25
3.3.2.3 Sample and Standard Preparation	26
3.3.2.4 Preparation of Blank	26
3.3.2.5 Experimental Procedure	26
3.3.3 Determination of Total Antioxidant Capacity (T	TAC)27
3.3.3.1 Reagents and Chemicals	27
3.3.3.2 Reagent Preparation	27
3.3.3.3 Sample and Standard Preparation	28
3.3.3.4 Preparation of the Blank	28
3.3.3.5 Experimental Procedure	28
3.4 Cytotoxicity Study by MTT Assay	29
3.4.1 N4X4 Cell Line	29
3.4.2 Solution Preparation	29
3.4.2.1 1% Penicillin–streptomycin Solution	29
3.4.2.2 10 % Fetal Bovine Serum	29
3.4.2.3 2% DMSO Solution	30
3 4 2 4 Trypsin	30

	3.4.2.5 Used Consumables	)
	3.4.2.6 Used Instruments	)
	3.4.2.7 Celltiter 96 Assay Kit	)
	3.4.3 Preparation of the Different Concentrations of Plant Extract	)
	3.4.4 Cell Culture	,
	3.4.4.1 Preparation of Assay Plates	,
	3.4.4.2 Thawing of Cells	,
	3.4.4.3 Cell Passage	,
	3.4.4.4 Harvesting of Cells	,
	3.4.4.5 Counting of Cells	,
	3.4.5 Procedure	<b>;</b>
Chap	ter 4	
Resul	ts34	
	4.1 Determination of Percentage Yield of the Plant Extract <i>Triphala</i>	Ļ
	4.1.1 Total Weight of Powdered Plant Material34	!
	4.1.2 Calculation of Percentage Yield of the Extract	!
	4.2 DPPH (1, 1-diphenyl-2-picryl hydroxyl) Free Radical Scavenging Assay36	<u>,</u>
	4.2.1 Table of DPPH free Radical Scavenging Activity of Ascorbic Acid, Bibhitaki, Amlaki, Haritaki and Different Ratios of Triphala36	
	4.2.2 Graph of Absorbance vs. Concentration of Ascorbic Acid, Bibhitaki, Amlaki, Haritaki and Different Ratios of Triphala in DPPH free radical scavenging assay	
	4.2.3 Table of Percentage of Inhibition of Ascorbic Acid, Bibhitaki, Amlaki, Haritaki and different ratios of Triphala	)
	4.2.4 Graph of % Inhibition of DPPH Free Radical by Ascorbic Acid, Bibhitaki, Amlaki,  Haritaki and Different Ratios of Triphala	?

Chapter 6	56
Chapter 5	
4.5.2 Cytotoxicity Test on N4X4 Cell Line using 2:1:1 (A:B:H) ratio	50
4.5.1 Cytotoxicity Test on N4X4 Cell Line using 1:1:1(A:B:H) ratio	48
4.5 In vitro Cytotoxicity Test of Triphala by MTT Assay on N4X4 (H Glioma) Cell Line	
4.4.2 Table of Total Antioxidant Capacity (TAC) in Triphala	47
4.4.1 Calibration curve of Ascorbic Acid (Standard) at 695 nm	46
4.4 Total Antioxidant Capacity Determination	46
4.3.2 Table of total phenolic content in Triphala	45
4.3.1 Calibration Curve of Gallic Acid at 765 nm	45
4.3 Total Phenolic Content (TPC) Determination	45
4.2.5 Determination of the IC <sub>50</sub> of Ascorbic Acid, Bibhitaki, Amlaki, H Different Ratios of Triphala	

## **List of Tables**

Table 1: Name of the traditional medicinal plants, their parts used in the
treatment of common diseases
Table 2: Name of the traditional medicinal plants used in Bangladesh
(Ghani, 2003a)
Table 3: Compounds isolated from medicinal plants with their mechanism
of action
Table 4: Previous studies on pharmacological activities of Triphala (Gupta, 2010)
Table 5: Reagents and chemicals required to determine DPPH scavenging
activity23
Table 7: Reagents and chemicals required to determine total antioxidant
activity27
Table 8: % yield of Amlaki, Bibhttaki and Haritaki extracts
Table 9: Table of DPPH Free Radical Scavenging Assay at 517 nm
(Absorbance vs. concentration)36
of ascorbic acid, Bibhitaki, Amlaki and Haritaki36
Table 10: Table of DPPH Free Radical Scavenging Assay at 517 nm
(Absorbance vs. concentration) of the
different ratios of Triphala

Table 11: Table of % Inhibition of DPPH Free Radical by Ascorbic Acid,
Bibhitaki, Amlaki and Haritaki
Table 12: Table of % Inhibition of DPPH Free Radical by different ratios of
Triphala41
Table 13: IC <sub>50</sub> values of Ascorbic Acid, Bibhitaki, Amlaki, Haritaki and
Different Ratios of Triphala
Table 13: Total phenolic content (TPC) of Bibhitaki is marked as gallic acid
equivalent (GAE)45
Table 14: Total antioxidant content (TAC) of Triphala is marked as
Ascorbic Acid equivalent (AAE)
Table 15: Survival and inhibition of N4X4 cell line in different
concentrations along with their IC50 value and average absorbance using the
1:1:1 (A:B:H) ratio
Table 16: Survival and inhibition of N4X4 cell line in different
concentrations along with their IC <sub>50</sub> value and average absorbance using
2:1:1 (A:B:H) ratio

# **List of Figures**

Figure 1: Organs affected by oxidative damage	5
Figure 2: Structures of polyphenols exerting antioxidant effect	6
Figure 3: Structure of MTT and colored formazan product (Riss et al.,	
2013)	10
Figure 4(a): Phyllanthus emblica or Emblica officinalis (Amlaki)	16
Figure 4(b): Terminalia bellirica (Bibhitaki)	16
Figure 4(c): Terminalia chebula (Haritaki)	16
Figure 7: Hemocytometer	32
Figure 8: Graph of Absorbance vs. Concentration (µg/ mL) for Ascorbic	
Acid, Bibhitaki, Amlaki and Haritaki in DPPH free radical scavenging	
assay	38
Figure 9: Graph of Absorbance vs. Concentration ( $\mu g/\ mL$ ) for different	
ratios of Triphala in DPPH free radical scavenging assay	39
Figure 10: Graph of % Inhibition of DPPH Free Radical by Ascorbic Acid,	
Bibhitaki, Amlaki and Haritaki	42
Figure 11: Graph of % Inhibition of DPPH Free Radical by different ratios	
of Triphala	43
Figure 12: Standard curve of standard gallic acid at 765 nm to determine the	
TPC in sample	45

Figure 13: Standard curve of ascorbic acid at 695 nm for determining TAC
value in methanol extract of Triphala
Figure 14: Cell viability of methanol extract of Triphala, 1:1:1(A:B:H)
ratio, at different extract concentrations of 0.0025 mg/ mL, 0.025mg/ mL,
0.25~mg/ mL, 2.5~mg/ mL, 25~mg/ mL and $2%$ DMSO as control
respectively after incubating 48 hours in N4X4 cell line
Figure 15: Cell viability of methanol extract of Triphala, 2:1:1 (A:B:H)
ratio, at different extract concentrations of 0.0025 mg/ mL, 0.025mg/ mL,
0.25~mg/ mL,2.5~mg/ mL,25~mg/ mL and $2%$ DMSO as control
respectively after incubating 48 hours in N4X4 cell line
Figure 16: Cytotoxic effect of methanol crude extract on N4X4 cell line

## **List of Acronyms**

WHO: World Health Organization

IARC: International Agency for Research on Cancer

A: Amlaki

B: Bibhitaki

H: Haritaki

AA: Ascorbic Acid

ROS: Reactive Oxygen Species

TPC: Total Phenolic Content

TAC: Total Antioxidant Capacity

DPPH: 1, 1-Diphenyl-2-Picryl Hydrazyl

FCR: Folin Ciocalteu Reagent

MET: Methanol Extract of *Triphala* 

GA: Gallic Acid

LAF: Laminar Air Flow

## Chapter 1

#### Introduction

### 1.1 Medicinal plants used for phytotherapy

World Health Organisation (WHO) defines medicinal plants as plants that holds therapeutic properties or compounds useful as therapeutics or those that produce metabolites that act as useful drugs (WHO 2008). Different parts of plants are used in the modern treatment and these different parts of the medicinal plants can be used as different dosage forms, such as decoctions, pastes, powders, juices, infusions etc. Examples of other famous medicines are colchicines from *Colchicum autumnale*, quinine from *Cinchona* spp. and caffeine from *Thea sinensis* (Burbage, 1981). Phytotherapy means the treatment using natural plant-derived medicines and in today's world plant-derived medicines have become a primary source of pharmaceuticals due to their characteristic pharmacological activity on human body and the most useful ones are flavonoids, phenolic compounds, and alkaloids which hardly contains toxins (Devendra, Srinivas, & Solmon, 2012). According to the World Health Organization (WHO), about 80 % of developing countries rely on traditional medicines for their primary health care. (Sofowora, Ogunbodede, & Onayade, 2013). Table 1 illustrates some name of the traditional medicinal plants along with their parts used in the treatment.

Table 1: Name of the traditional medicinal plants, their parts used in the treatment of common diseases

Scientific Name (Family)	Part used	Traditional uses	References
Curcuma longa L	Rhizome	Irritation & itching, inflammation, erythema, burns & sunburn and other skin diseases	Zari, S. T., &Zari, T.A. (2015).
Plumbago zeylanica L.	Root	Paralysis, secondary syphilis, leprosy and ophthalmics	Chopra, R.N., Nayer, & Chopra, I.C. (1992).
Aloe vera L.	Leaf gel	Psoriasis, cold sores, sun burns, microbial infections on skin	Zari, S. T., &Zari, T.A. (2015).
Acanthus ilicifolius L	Root	Leucorrhoea	Hossan et al. (2010).
Holarrhena antidysenterica Wall	Bark	Dysentery, dropsy, fever, diarrhea and intestinal worm infections	Chopra, R.N., Nayer, & Chopra, I.C. (1992).

## 1.2 Overview of Bangladeshi Medicinal Plants

According to the WHO, about 80 % of population from developing countries rely on Bangladeshi medicinal plants for primary treatment. Furthermore, due to better cultural acceptability, improved compatibility with human body and fewer side effects about 25 % of prescribed drugs come from plants and herbal medicines (Jafor & Obaidullah, 2018). For many years numerous plants have been used to treat health conditions using traditional medicine. Some examples are given in Table 2.

Table 2: Name of the traditional medicinal plants used in Bangladesh (Ghani, 2003a)

Bangali	Scientific	Parts of	Uses
Name	Name	plants	
Mehedi	Lawsonia inermis	Leaves, flower	<ol> <li>Skin disease</li> <li>Pox</li> <li>Burn</li> <li>Dandruff</li> </ol>
Thankuni	Cliotoria ternatea	Whole plant	<ol> <li>Weakness</li> <li>Dermatitis</li> <li>Jaundice</li> <li>Stomach disorder</li> </ol>
Arjun	Terminalia arjuna	Bark	1. Heart disease
Halud	Curcuma longa	Rhizomes	<ol> <li>Blood purification</li> <li>Skin and eye disease</li> <li>Stomachache</li> </ol>

Continued

Lajjabati	Mimosa pudica	Whole plant	1. Toothache
			2. Convulsion
			3. Blood
			purification
			4. Fistula
			5. Piles
			1. Cancer
Nayantara	Catharanthus roseus	Flowers	2. Diabetes
			3. Insomnia
			4. Blood pressure

## 1.3 Reactive Oxygen Species: Source and Side-effects

Aerobic metabolism in the cell is continuously making oxygen-derived species such as hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>), and hypochlorous acid (HOCl) and free radicals such as superoxide (O<sub>2</sub>), nitric oxide (NO) and hydroxyl (OH). Furthermore, exposure to oxidizing air pollutants, tobacco smoke, and car exhausts lead to the production as well. Even with the body's antioxidant defense systems to protect against these reactive oxygen species, oxidative damage is imminent. Severe stress can damage DNA, proteins, and lipid causing cell transformation or cell death, thus causing human diseases such as cardiovascular disease and cancer (Helliwell B. and Aruoma, 1993, Helliwell B., 1994, Halliwell B, Gutteridge, 1989).

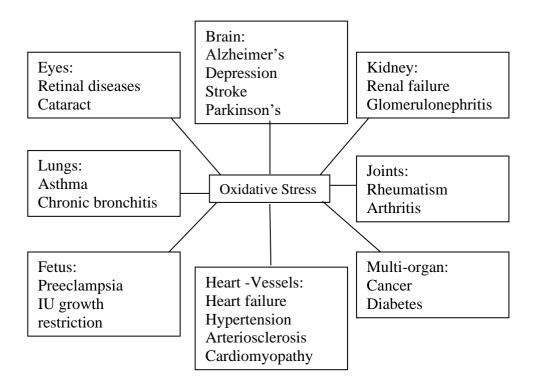


Figure 1: Organs affected by oxidative damage

In the figure 1, it is highlighted that the buildup of oxidative stress caused by reactive oxygen species (ROS) in several organs will destroy the cells and eventually cause diseases.

#### 1.4 Antioxidants

Antioxidants are compounds that considerably reduces the oxidation of a substrate at small concentrations compared to that of an oxidizable substrate (Young, 2001). Natural as well as synthetic antioxidants are used to treat diseases and prevent cellular damage done by ROS or free radicals. The medicinal plants exhibiting antioxidant properties generally contain polyphenols and flavonoid, a polyphenolic chemical, which are considered the main antioxidant compounds.

#### 1.4.1 Significance of Antioxidant Potential of Plants

Plants have an intrinsic ability to generate a wide range of non-enzymatic cancer prevention agents which work to mitigate the oxidative damage stimulated by ROS. Plants promote the production and aggregation of low-weight cancer preventive agents such as

vitamin C, vitamin E, phenolic acids, and high-weight cancer preventive agents such as tannins. These work as reducing agents, free radical scavengers and metal chelators. To prevent the toxic consequences of the free radicals,

plants have effective enzymatic defence systems called catalase (CAT), glutathione peroxidase(GPx), and glutathione reductase (GR) and non-enzymatic defense systems called ascorbic acid, glutathione, lignan, pro-line, stilbene, carotenoids, phenolic acids, flavonoids, tannins etc. Polyphenols and carotenoids are the two major classes of natural antioxidants in our foods and medications. (Tsao, 2010).

Figure 2: Structures of polyphenols exerting antioxidant effect

Lignan

Tannin

Flavonoids

Phenolic acids

#### 1.5 *In vitro* Antioxidant Assay

The use of DPPH, TAC, TPC and TFC assays can determine radical scavenging capacity of various extracts. Compared with other available methods, these methods are more reliable, accurate and cheap. (Thaipong, Boonprakob, Crosby, Cisneros-Zevallos & Hawkins Byrne, 2006). When compared with other procedures for free radical scavenging, DPPH is considered the fastest and most convenient (Alam et al, 2013).

The reaction between DPPH and antioxidants (H-A) is given below:

#### DPPH + HA: DPPH-H + HA

In the reaction, after the substrate reacts, it turns DPPH (1, 1-diphenyl-2-picrylhydrazyl) yellow from red color. Antioxidants bind to form DPPH-H with the loss of red color and ultimately leading to diminished absorbance. This discoloration indicates antioxidant compound's scavenging activity.

#### 1.6 Brain Tumour

Brain tumor refers to a mixed group of neoplasms that form from the intracranial tissues and abnormal brain cell growth due to mutations. Brain tumors are the second most common cancer in children, and consist around 15-25% of pediatric malignancies, around 40% glioma and around 25% medulloblastomas.

#### 1.6.1 Types of Glioma

The main types of gliomas are Astrocytomas, Oligodendoglioma, Ependymomas and Glioblastoma.

#### 1.6.2 Anticancer drugs from Medicinal Plants

For their basic health needs, about 80 per cent of the total population of third world nations relies solely on plant products (Mans et al., 2000). Researchers noted that herbal medicinal products are considered to be one of the best treatments for cancer and some known

metabolites with anticancer effects are flavonoids, terpenoids, alkaloids, and phenylpropanoids (Kumar et al., 2012). These compounds may be used alone or in combination with other anticancer treatments. These natural compounds, compared to synthetic drugs, are more easily available, cheaper and can be easily administered orally and at the same time have negligible side effects (Seca & Pinto, 2018; Lichota & Gwozdzinski, 2018). Thazin and colleagues mentioned that natural plant compounds are capable of working as both an anticancer agent as well as for restoring sensitivity to chemotherapy. For example, doxorubicin anticancer activity towards resistant MCF-1/DOX cells in vivo is increased by tetrandrine and it is an active alkaloid compound extracted from plants (Aung et al., 2017). The first agents that have been discovered as an anti-cancer agent are vinca alkaloids (Vinblastine and Vincristine) from the periwinkle plant Catharanthus roseus. Some examples of anticancer compounds isolated from plants is shown in Table 3.

Table 3: Compounds isolated from medicinal plants with their mechanism of action

Compound	Drug Class	Plant source	Mechanism of action
Vinblastin,		Catharanthus	Polymerization
vincristine	Vinca alkaloids	roseus	inhibition of tubulin
D ( 1	T	T 1	Microtubule
Docetaxel	Taxanes	Taxus baccata	stabilization
Berbamine	Bis-benzylisquinoline	Berberis amurensis	Appoptosis
Hypericin	Nipthodianthrone	Hypricum perforatum	Hypercinoids inhibition
Etoposide,		Catharanthus	Topoisomerase
Tenoposide	Alkaloids	roseus	inhibitor
Irinotecan,		Catharanthus	Topoisomerase
Topotecan	Alkaloids	acuminate	inhibitor

Continued

### 1.7 In vitro Cytotoxicity Screening

Cytotoxicity refers to the property of a chemical capable of causing cell death. The introduction of a cytotoxic agent to a normal or cancerous cell will induce necrosis of the cells and if it kills both normal and cancerous cells, then it is a cytotoxic agent, but if it kills only cancerous cell then it can be a prospective anticancer agent. Different methods for evaluating cytotoxic activity have been developed (Adan, Kiraz & Baran) and MTT assay has been shown to be sensitive, quantitative and accurate among all the tests (Senthilraja & Kathiresan, 2015).

### 1.7.1 MTT (3 - (4,5 – dimethylthiazol-2-yl) - 2, 5 - diphenyltetrazolium bromide) assay

MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide assay is a type of tetrazolium reduction assay that iis commonly used to find viability of the cells. First MTT substrate is prepared in a solution, then transferred to cultivated cells, and eventually incubated for 1 to 4 hours. The hypothesis shows the amount of formazan generated is proportional to the number of cells that are viable. The measurement is done using a plate reading spectrophotometer where absorbance is taken at 570nm. (Riss et al.,2013). The reaction is shown in Figure 3.

Figure 3: Structure of MTT and colored formazan product (Riss et al., 2013).

### 1.8 Selection of *Triphala* for the Study

Triphala ['three' (tri) 'fruits' (phala)] is a traditional Ayurvedic herbal formulation consisting of medicinal plant fruits, *Terminalia chebula* (Haritaki), *Terminalia bellirica* (Bibhitaki), and *Phyllanthus emblica* or *Emblica officinalis* (Amlaki). Antioxidant capacity determination of methanol extract was not carried out before in non-equivalent ratios. Methanol extract of *Triphala* showed the highest antioxidant activity among all the solvents. *Triphala* also showed cytotoxic and apoptosis activity against different cancer cell lines, but no study was done using N4X4 cell lines. That is why, methanol extract of *Triphala* is selected for the antioxidant and cytotoxic activity determination with N4X4 cell lines in 1:1:1 (A:B:H) ratio and 2:1:1 (A:B:H) ratio.

### 1.9 Rationale of This Study

Triphala possess free radical scavenging, antioxidant, anti-inflammatory, antipyretic, analgesic, antibacterial, antimutagenic, wound healing, anticariogenic, adaptogenic, hypoglycaemic, anticancer, chemoprotective, radioprotective and chemopreventive effects. However different ratios of Triphala as a methanol extract was not used before for the assessment of antioxidant activity. In addition, methanol extract of this plant showed strong cytotoxicity against other cancer-cell lines but N4X4 cell lines were not used before to determine the cytotoxic potential. Moreover, non-equivalent ratio of

Triphala was not used to do antioxidant and cytotoxicity test.

### 1.10 Aim of This Study

The aim of this project is to investigate the *in vitro* antioxidant and cytotoxic activity of methanol extract of *Triphala* (*Terminalia chebula* (Haritaki), *Terminalia bellirica* (Bibhitaki), and *Phyllanthus emblica* or *Emblica officinalis* (Amlaki))

#### 1.11 Objectives of This Study

After studying the literature review from previous findings on Triphala, objectives were made as follows:

- Determination of antioxidant activity using various *in vitro* methods and assessment of free radical scavenging potential comparable to that standard ascorbic acid using the DPPH free radical scavenging method and others
- Determination of cytotoxic activity using MTT assay on N4X4 cell line.

#### 1.12 Introduction of the Plant

Triphala ['three' (tri) 'fruits' (phala)] is a formulation consisting of *Phyllanthus emblica* or *Emblica officinalis* (Amlaki), *Terminalia chebula* (Haritaki), and *Terminalia bellirica* (Bibhitaki).

## 1.12.1 Phyllanthus emblica or Emblica officinalis (Amlaki)

#### 1.12.1.1 General Description

Amlaki can be found in India, around the sea-coast districts and on hill slopes of up to 200 meters (Calixto, Santos, Filho & Yunes, 1998) and also found in Burma (Dey KL, 1896). The Emblica officinalis is a medium to small—sized deciduous tree, 8 18 meters high with a thin light grey bark, the leaves are flat, light green with pinnate 1 eaves. The flowers are greenish yellow, in axillary fascicles, unisexual (Singh and Singh, 2002). Figure 4(a) shows the picture of the fruit.

1.12.1.2 Traditional Uses

Amlaki is traditionally used for many things such as to keep control of blood sugar

level, decreasing total cholesterol, eye diseases like cataract and glaucoma. It also has

nourishing effect on hair when used as a hair tonic and also on the skin. It is also used

for hyperacidity, as a mixture with sugar syrup. (Kumar, 2012)

1.12.1.3 Constituents

EO primarily contains tannins, alkaloids, phenolic compounds, amino acids and

carbohydrates. Its fruit juice contains the highest vitamin C (478.56 mg/100 mL). The

fruit hincreases nutritional quality in terms of Vitmin C when mixed with other fruits

(Shastri and Bhavaprakasha, 1969). Compounds isolated from EO were gallic acid,

ellagic acid, 1-O-galloyl-beta-Dglucose, 3, 6-di-O-galloyl-Dglucose, chebulinic acid,

quercetin, chebulagic acid, corilagin, 1, 6-di-O - galloyl beta D glucose, 3 Ethylgallic

acid (3 ethoxy 4, 5 dihydroxy benzoic acid) and isostrictiniin21. Phyllanthus emblica

also contains flavonoids, kaempferol 3 O alpha L (6" methyl) rhamnopyranoside and

kaempferol 3 O alpha L (6"ethyl) rhamnopyranoside22. A new acylated apigenin

glucoside (apigenin 7 O (6" butyryl beta glucopyranoside) was isolated from the

methanolic extract of the leaves of Phyllanthus emblica (Nadkarni, 1993).

1.12.1.4 Taxanomical Classification

Kingdom: Plantae

Division:

Angiospermae

Class:

Dicotyledonae

Order:

Geraniales

Family:

Euphorbiaceae

Genus:

Emblica

Species:

Officinalis

1.12.2 Terminalia bellirica (Bibhitaki)

1.12.2.1 General Description

It is found in the wild throughout the India, Srilanka, and Southeast Asia. It can be seen

at 1200 meters in elevation, in a wide variety of ecologies. It is a large deciduous tree

with a buttressed trunk, a thick brownish gray bark, attaining a height of between 20 and

30 meters. The leaves are crowded around the ends of the branches, alternately arranged,

margins entire, elliptic to ellipticobovate. The flowers are pale greenish yellow with an

offensive odor. The fruits are ovoid grey drupes. (Saroya, 2009, Nadkarni, 2002). Figure

4(b) shows the picture of the fruit.

1.12.2.2 Traditional Uses

Bibhitaki is traditionally used for common cold, pharyngitis and constipation. The bark

is midly diuretic and is useful in anaemia and leucoderma. Unripe fruit is a mild laxative

and ripe fruit is an astringent. The seeds are used as aphrodisiac and the oil extract from

the seed pulp is used in leucoderma and alopecia. (Kumudhavalli, 2010).

1.12.2.3 Constituents

Glucoside (bellericanin), Gallo-tannic acid, Coloring matter, resins and a greenish

yellow oil. Ellargic acid, gallic acid, lignans (termilignan and thanni lignan), 7-hydroxy

3'4' (methylene dioxy) flavone and anolignan B10. Tannins, ellargic acid, ethyl gallate,

galloyl glucose and chebulaginic acid, phenyllemblin, β-sitosterol, mannitol, glucose,

fructose and rhamnose (Saroya, 2009, Singh, 2006).

1.12.2.4 Taxonomical classification

Kingdom: Plantae

Division: Magnoliophyta

Class:

Magnoliopsida

Order:

Myrtales

Family:

Combretaceae

Genus:

Terminalia

Species:

bellirica

1.12.3 Terminalia chebula (Haritaki)

1.12.3.1 General Description

T. chebula is native in Asia and mainly found in Nepal, Sri Lanka, Myanmar,

Bangladesh, Egypt, Iran, Turkey, Pakistan, Yunnan, Tibet, Guangdong, Guangxi

province of China. It is a medium-sized, up to 25 m tall, deciduous tree, bark dark brown.

Leaves alternate or opposite, thin-coriaceous, ovate or elliptic-obovate, petiole up to 2

cm long. Flowers in axillary 5-7 cm long spikes, simple or sometime branched,

yellowish-white and unpleasantly scented. Fruit an obovoid or oblong-ellipsoid drupe,

yellow to orange-brown when ripe. (Boer et al., 1995, Fundter et al., 1992, Prakash,

2012). Figure 4(c) shows the picture of the fruit.

1.12.3.2 Traditional Uses

Haritaki is used as an eye tonic, mainly for eye irritations, and also as a hair tonic to

reduce greying of hair. It also helps to control vomiting, diarrhea, can increase appetite

and also works as a digestive. A liver tonic is also made, which is a jam, for liver

problems and also used as a diuretic and treatment of polypus (Chandra, 2012).

1.12.3.3 Constituents

In Terminalia chebula, 33% of the total phytoconstituents are hydrolysable tannins and

are responsible for pharmacological activity. These tannins contain phenolic carboxylic

acid like gallic acid, ellagic acid, chebulic acid and gallotannins such as 1,6 di-O-galloyl-

β-D-glucose, 3,4,6 tri-O-galloyl-βD-glucose, 2,3,4,6 tetra-O-galloyl-β-D-glucose,

1,2,3,4,6 penta-Ogalloyl-β-D-glucose. Ellagitannin such as punacalagin, casurarinin,

corilagin and terchebulin and others such as chebulanin, neochebulinic acid, chebulagic

acid and chebulinic acid reported in literature (Juang et al., 2004, Han et al., 2006). The

tannin content varies with the geological variation. Flavonol glycosides, triterpenoids,

coumarin conjugated with gallic acid called chebulin, as well as phenolic compounds

were also isolated. (Chottopadhyay and Bhattacharyya, 2006)

#### 1.12.3.4 Taxonomical classification

Kingdom: Plantae

Division: Magnoliophyta

Class: Magnoliopsida

Order: Myrtales

Family: Combretaceae

Genus: Terminalia

Species: chebula



Figure 4(a): Phyllanthus emblica or Emblica officinalis (Amlaki)

("Triphala", 2020)



Figure 4(b): Terminalia bellirica (Bibhitaki) ("Triphala", 2020)



Figure 4(c): Terminalia chebula (Haritaki) ("Triphala", 2020)

## Chapter 2 Literature Review

## 2.1 Previously Studied Pharmacological Activities of Triphala

Literature review of *Triphala* and its constituents has been done and it showed that they carry different types of activities as illustrated in table 4.

Table 4: Previous studies on pharmacological activities of Triphala (Gupta, 2010)

Use	Amlaki	Haritaki	Bibhitaki
External Use	<ul><li>Hair fall</li><li>eye diseases</li><li>various skin</li><li>diseases</li></ul>	<ul><li>Pain</li><li>Inflammation</li><li>Wounds</li><li>various skin diseases</li></ul>	<ul> <li>Pain</li> <li>Leukoderma</li> <li>Inflammation</li> <li>greying of</li> <li>hair</li> </ul>
Central Nervous System	<ul> <li>Diseases of sense organs</li> <li>Enhance memory</li> </ul>	<ul> <li>Various <ul> <li>neurological</li> <li>disorders</li> </ul> </li> <li>Enhance memory</li> <li>Eye disease</li> </ul>	<ul> <li>Insomnia and         various central         nervous system         (CNS) disorders</li> <li>Prevents         blindness, myopia         and eye disorders</li> </ul>
Cardio- vascular System	<ul><li>Cardiac diseases</li><li>Blood Disorders</li></ul>	<ul><li>Cardiac diseases</li><li>Oedema</li><li>Anaemia and blood discracias</li></ul>	<ul><li>Bleeding disorders</li><li>Haemoptysis</li></ul>
GIT	• Anorexia	• Anorexia	Anorexia

17 Continued

	• Impaired	Colic pain	• Upward
	digestion	Haemorrhoids	movement of
	Constipation	• Jaundice	gases
	Liver disorder	Diseases of liver	Vomiting
	• Gastritis	and spleen	Constipation
	Colic pain	Helmentiasis	Haemorrhoids
	Haemorrhoids	Abdominal	Helmentiasis
	• Other	tumours	
	gastrointestinal	Abdominal	
	disorders	ailments	
	• Cough,	• Rhinitis	
	difficulty in	• Cough	
Respiratory	breathing	Harshness of	
1 - 1	Tuberculosis	voice	
	• Various	• Bronchitis	
	respiratory	Other respiratory	
	disorders	disorders	
		Oligurea	
Genito- urinary System	Oligurea	Renal calculi	Aphrodisiac
	• Dysuria •	Urinary disorders	Improves libido
	Aphrodisiac	• Gout	and impotence
	Infertility	Aphrodisiac	and imposence
		• Infertility	

# 2.1.1 Triphala as An Antineoplastic Agent

In last fifty years, chemotherapy has been the foundation of cancer treatment. Most chemotherapeutic medications, however, have toxicity and this has led to the need for an accurate, readily acceptable and inexpensive non-toxic agent. Preclinical studies on Michigan Cancer Foundation-7 (MCF-7) and T47D, which are human breast cancer cells, have shown that Triphala has antineoplastic effects (Kaur et al., 2005; Sandhya et al., 2006; Sandhya & Misha, 2006). Furthermore, it showed effect on PC-3 and DU-145, which are prostate cancer cells (Kaur et al., 2005), Capan-2, BxPC-3 and HPDE-6 cells, which are human pancreatic cancer cells (Shi et al., 20008), Shionogi 115, which are mouse breast cancer cells (Kaur et al., 2005) and barcl-95 transplantable mouse thymic lymphoma (Sandhya et al., 2006) in vitro and repressed the development of Capan-2 (pancreatic tumor) in nude mice (Shi et al., 20008). Research have also shown that Triphala has no cytotoxic effects on normal breast epithelial cells, mouse liver and spleen cells, human peripheral mononuclear blood cells (Sandhya et al., 2006), and HPDE-6 cells (Shi et al., 20008). Studies show that Triphala regulated the cytotoxic effects by increasing intracellular ROS and primarily by inducing apoptosis by the p53 pathway (Sandhya et al., 2006; Sandhya & Misha, 2006; Shi et al., 20008).

#### 2.1.2 Triphala as A Radioprotective Effect

Radiotherapy, while effective in treating cancer causes damage to healthy tissues. The use of radioprotectors that can systemically shield the normal cells from damage is a possible strategy for increasing the therapeutic index, thereby offering tremendous benefit to the patients with cancers. Preclinical studies have shown that Triphala has radioprotective properties when delivered through intraperitoneal (Jagetia et. al, 2002; Jagetia et al.; 2004) and oral routes. When delivered intraperitoneally, 10 mg / kg Triphala was found to be most successful if administered before the irradiation and

caused a reduction in dose. Biochemical tests found that Triphala stabilized levels of xanthine oxidase (XO) and SOD activity in mice's intestine and also minimized damage of DNA in blood leukocytes and splenocytes, suggesting there was inhibition of oxidative damage in cells and organs. (Sandhya et al., 2006)

# 2.1.3 Triphala as A Chemoprotective Agent

The antifolate class of anticancer drug methotrexate causes severe enterotoxicity which sometimes reduces the therapeutic benefit. Nariya et al researched 1:1:1 and 1:2:4 triphala ratio for Enteroprotective results against methotrexate-induced damage on the intestines of rats by evaluating gross and microscopic harm, testing intestinal permeability to red phenol and biochemical tissue parameters. The authors found that Triphala administration regained the reduced protein level in the brush borders of the intestine, phospholipid, and glutathione amount and decreased Myeloperoxidase (MPO) and XO levels in the mucosa of methotrexate-treated rat's inestine, with greater effects being observed with 1:2:4 mixture. (Nariya et al., 2009)

#### 2.1.4 Triphala as A Chemopreventive Agent

Chemoprevention are non-toxic organic or naturally occurring chemicals, that prevent cancer or reduce the incidence of cancer. Deep, et al showed that Triphala decreased the B(a) P-induced forestomach papillomagenesis in mice and the result depended both on concentration and time. Triphala was shown to be more effective at equal concentration than the individual constituents suggesting that the three constituents act synergestically to mediate the observed chemopreventive effects. (Deep et al., 2005)

#### 2.1.5 Antimicrobial Activity

Reportedly, Triphala and its individual constituents have strong antibacterial activity on different organisms. Triphala's aqueous and ethanol extracts, as well as the

separate components, have shown antibacterial efficacy against isolates acquired from patients diagnosed with the human immunodeficiency virus (HIV) (Srikumar et al., 2007) Furthermore, the ethanolic and aqueous extracts have shown antibacterial activity against the antibiotic resistant strains of bacteria and the aqueous extracts were more effective than ethanolic extracts. (Biradar et al., 2008)

#### 2.1.6 Antihyperglycemic Effects

Today, diabetes affects about 5 percent of the global population and is the leading endocrine disease in the world. Animal studies show that the delivery of Triphala's equivalent concentration and the individual components decreased serum glucose concentrations in both usual and diabetic rats nduced by alloxane. (Sabu & Kuttan, 2007)

#### 2.1.7 Antipyretic and Analgesic Activities

In a study the yeast-induced pyrexia in mice had decreased after administration of triphala, similar to indomethacin, which is the positive control. Triphala has also shown analgesic effect on mice which had acetic acid induced writhing and hot plate assay. (Sabina & Rasool, 2007)

#### 2.1.8 Antioxidant Effects

The eukaryotic cells are loaded with natural antioxidant molecules like glutathione(GSH), vitamin E, vitamin A, carotenoids, vitamin C, etc. and antioxidant superoxide dismutase (SOD), catalase (CAT), glutathione peroxidase(GPx) to protect the cell from free radical-induced harm. Animal studies have shown that the administration of Triphala has raised SOD, GPx and CAT activities and increased GSH levels, that caused significant decline in the gastric tumors in mice.(Deep al., 2005) Studies also show that Triphala et reduced myeloperoxidase(MPO) and xanthine oxidase(XO) amounts in the intestinal mucosa of methotrexate-treated rats with a simultaneous increase in the levels of GSH (Nariya et al., 2009)

# Chapter 3

# Methodology

# 3.1 Collection and Authentication

After investigation and having reviewed many literatures, it was strongly observed that many individual and combined studies have been conducted on these three fruits. Studies have shown that methanol has exposed the highest activity for antioxidants and against cancer cell line which is why methanol was chosen as the solvent. Fortunately, antioxidant activity of methanol extract in different ratios and cytotoxic activity against glioma (N4X4) cell line in 1:1:1 (A:B:H) and 2:1:1 (A:B:H) ratio of mixture was not used. The plant powders were obtained from commercially available packets.

#### 3.2 Extraction Procedure

#### 3.2.1 Extraction

The maceration process was used for extraction of plant materials and methanol was used as the organic solvent. 600 gm of the coarse powder of Amlaki, Haritaki and Bibhitaki was taken separately in three fresh, round bottomed flask (5 liters) and soaked in 1.2 liter of methanol each. With occasional agitation, the three materials in the flasks were soaked in methanol at normal room temperature (22-25°C) for a period of 7 days.

#### 3.2.2 Filtration

The contents of the beaker were first decanted, after seven days of maceration, while being filtered using fresh cotton fabric, followed by cotton plug and Whitman filter (pore size: 110 mm).

#### 3.2.3 Concentration and Drying

Using a rotary evaporator (Heidolph), the collected filtrate was concentrated at 100 rpm at 30°C until concentrated methanol extract was produced for each of the three extracts. The mixture was then transferred to a petri dish and placed under Laminar Air Flow (LAF) to evaporate the extract solvent, leaving the extract dry and semi-solid. LAF is also a form of prevention for microbial growth on the extract while it is drying. The petri dishes were covered in aluminum foil and labeled after successful drying and kept in a cool location.

# 3.3 In vitro Antioxidant Assay

There are various *in vitro* methods for estimation of antioxidant activities of plant extracts. DPPH free radical scavenging assay, total phenolic content (TPC) and total antioxidant capacity (TAC) methods were selected among the other methods to estimate the antioxidant activity in methanol extract of Triphala.

#### 3.3.1 DPPH (1, 1-diphenyl-2-picryl hydrazyl) Free Radical Scavenging Assay

The DPPH free radical scavenging assay of methanol extract of Triphala was done using the method described by (Lalhmingmawii & Jagetia, 2018)

#### 3.3.1.1 Reagents and Chemicals

The reagents and chemicals for the determination of DPPH scavenging activity are listed below in Table 5.

Table 5: Reagents and chemicals required to determine DPPH scavenging activity

Name of reagent/chemical	Source
DPPH	Sigma Aldrich U.S. A
Methanol	Active Fine Chemicals Ltd, Bangladesh
L ascorbic acid	Merck, Germany

#### **3.3.1.2 Sample and Standard Solution Preparation**

120 mg of extract was measured and dissolved in 10 mL of ethanol to produce a concentration of 12 mg/ mL. This became the stock solution and further concentrations were prepared by dilution of the sample stock solution to derive 8 concentrations: 1200, 800, 400, 200, 100, 50, and 25  $\mu$ g/ mL. The standard used was L-ascorbic acid and it was prepared similarly to get 8 eight diluted concentrations, ranging from 1200-12.5  $\mu$ g/ mL.

# 3.3.1.3 DPPH solution preparation

4 mg of DPPH was weighed and dissolved in 100 mL methanol to get 0.004% w/v of DPPH solution. This was then stored in the refrigerator at -4°C till before use.

#### **3.3.1.4** Experimental Procedure

- a. 1 mL of each of the fractions of sample and standard (L-ascorbic acid) were taken in test tubes.
- b. To each of the test tubes, 2 mL of 0.004 % (w/v) DPPH solution was added.
- c. Control was prepared by adding 1 mL methanol and 2 mL DPPH.
- d. Then, the test tubes were incubated for 30 minutes at room temperature which was followed by measuring the absorbance of the resulting solutions and control at 517 nm against blank (methanol) using a spectrophotometer (U- 2910 UV-Vis Spectrophotometer).
- e. The percentage of free radical scavenging activity (% FRS) was then calculated from the equation as follows:

% Free radical Scavenging (% FRS) activity = 
$$\underbrace{(A0-A1)}_{A0} \times 100$$

Where,  $A_0$  = absorbance of the control  $A_1$  = absorbance of the sample/standard

- f. Finally, the % Scavenging activity was plotted against concentration from which IC<sub>50</sub> value was calculated.
- g. IC<sub>50</sub> value is defined as the concentration at which 50 % of total DPPH free radical is reduced by the antioxidants which can be determined by plotting inhibition % against the respective concentrations.

#### 3.3.2 Determination of Total Phenolic Content (TPC)

The TPC of the plant extract of mixture was determined by the modified Folin-Ciocalteu method (Chandra et al., 2014).

#### 3.3.2.1 Reagents and Chemicals

The reagents and chemicals for the determination of phenolic content are listed in Table 6.

Table 6: Reagents and chemicals required to determine flavonoid content

Name of reagent/chemical	Source
Folin-Ciocalteu reagent (FCR)	LOBA Chemicals Pvt. Ltd., India
Gallic Acid Monohydrate (Standard)	Sigma Aldrich, USA
Sodium Carbonate	Merck Specialties Pvt. Ltd., Mumbai
Methanol	Active Fine Chemicals Ltd., Bangladesh

#### 3.3.2.2 Reagent Preparation

100 mL of 10 % Folin-Ciocalteu reagent (FCR) solution was prepared by taking 10 mL of FCR in a volumetric flask and diluting it with distilled water. 100 mL of 7.5 % (w/v) Sodium carbonate was prepared by measuring out 7.5 g of sodium carbonate in a 100 mL volumetric flask and then diluting it with distilled water.

#### 3.3.2.3 Sample and Standard Preparation

120 mg of extract was measured and dissolved in 10 mL of ethanol to produce a concentration of 12 mg/ mL. This became the stock solution and further concentrations were prepared by dilution of the sample stock solution to get four diluted concentrations: 1200, 800, 400 and 200  $\mu$ g/ mL. Gallic acid was used as the standard and was prepared in the same manner as the extract.

#### 3.3.2.4 Preparation of Blank

The blank solution contained 5 mL FCR solution and 4 mL sodium carbonate and 1 mL of ethanol was used to make the volume up to 10 mL.

#### 3.3.2.5 Experimental Procedure

- a. 0.5 mL of each of the fractions of sample and standard were taken in test tubes.
- b. To each of the test tubes, 2.5 mL of FCR was added.
- c. 2 mL of sodium carbonate solution was added.
- d. Each of the mixtures was vortexed for 15 s and then allowed to stand for 30 min at  $40^{\circ}$ C in a water bath.
- e. Finally, the absorbance of standard and sample solutions was measured against blank at 765 nm using spectrophotometer (U-2910 UV-Vis Spectrophotometer).
- f. The total phenolic content, C, for each of the fractions were expressed as Gallic Acid Equivalents (GAE) using the following equation:

$$C = \frac{c \times V}{m}$$

Where, C = Total content of phenolic compounds, milligram of Gallic acid per gram of dried plant extract, expressed as Gallic acid equivalents (QE)

Gallic Acid Equivalent (GAE) c = Concentration of Gallic acid obtained from calibration curve (mg/ mL)

V = Volume of sample solution (mL), m = Weight of the sample (g)

# **3.3.3 Determination of Total Antioxidant Capacity (TAC)**

The total Antioxidant capacity of plant extract of *Triphala* was determined by using the following method described by (Pisoschi & Negulescu, 2012).

# 3.3.3.1 Reagents and Chemicals

The reagents and chemicals for the determination of total antioxidant capacity are listed below in Table 7.

Table 7: Reagents and chemicals required to determine total antioxidant activity

Name of reagent/chemical	Source
Ammonium Molybdate	Active Fine Chemicals Ltd., Bangladesh
L-ascorbic acid	Merck, Germany
Trisodium Phosphate (Na <sub>3</sub> PO <sub>4</sub> .12H <sub>2</sub> O)	Merck, Germany
Methanol	Active Fine Chemicals Ltd., Bangladesh
Concentrated Sulfuric Acid	Merck, Germany

#### 3.3.3.2 Reagent Preparation

50 mL of 0.6 M Sulfuric acid was prepared by measuring 2.64 mL of 98 % concentrated sulfuric acid in a 50 mL volumetric flask and diluting it with distilled water. 50 mL of ammonium molybdate solution was prepared by measuring 2.247 g of ammonium molybdate in a 50 mL volumetric flask and diluting it with distilled water. 50 mL of 0.028 M trisodium phosphate solution was prepared by measuring 0.532 g of trisodium phosphate in a 50 mL volumetric flask and diluting it with distilled water.

#### 3.3.3 Sample and Standard Preparation

120 mg of extract was measured and dissolved in 10 mL of ethanol to produce a concentration of 12 mg/ mL. This became the stock solution and further concentrations were prepared by dilution of the sample stock solution to get four diluted concentrations: 1200, 800, 400 and 200  $\mu$ g/ mL. Ascorbic acid was used as the standard and the stock solution was prepared in the same manner as the extract resulting in four serially diluted concentrations, ranging from 1200, 800, 400 and 200  $\mu$ g/ mL.

# 3.3.4 Preparation of the Blank

3 mL of reagent solution and 300  $\mu$ L of methanol was used for preparing the blank solution.

#### 3.3.3.5 Experimental Procedure

- a. 300 µL of each of the fractions of sample and standard (L-ascorbic acid) concentrations were taken in test tubes.
- b. 3 mL of Reagent solution (0.6 M sulfuric acid, 0.028 M sodium phosphate and 0.004
   M ammonium molybdate) was added into the test tubes.
- c. The test tubes (sample, standard and blank) were then incubated at 95°C in a water bath for 90 min.
- d. Finally, the absorbance of the sample and standard solutions was measured against blank at 695 nm using spectrophotometer (U-2910 UV-Vis Spectrophotometer).
- e. The total antioxidant capacity, A, for each of the fractions were expressed as:

$$C = \frac{c \times V}{m}$$

Where, C = Total antioxidant capacity, milligram of ascorbic acid per gram of dried plant extract, expressed as ascorbic acid equivalent (AAE)

c = Concentration of ascorbic acid obtained from calibration curve (mg/mL)

V = Volume of sample (mL), m = Weight of the sample (g)

# 3.4 Cytotoxicity Study by MTT Assay

#### **3.4.1 N4X4 Cell Line**

N4X4 cell line is a glioma NP-2 cell line with stable expression of CXCR4 produced in the  $M_{10}$  medium (MMM; Sigma, St. Louis, MO, USA) with a 10 % FBS supplement, 100 units/ mL penicillin and 0.1 mg/ mL streptomycin).

# **3.4.2 Solution Preparation**

#### 3.4.2.1 1% Penicillin-streptomycin Solution

Penicillin-streptomycin solution usually known as pen-strep is used in MTT assay to control the bacterial contamination and maintain the sterile condition throughout the process. The solution contains a mixture of 10000 units of penicillin per mL and 10 mg of streptomycin per mL. In case of N4X4 cell line, RPMI containing 1% penicillin-streptomycin is used where RPMI 1, also known as the RPMI medium, is a cell culture growth medium. In case of N4X4 cell line, DMEM containing 1% penicillin-streptomycin is used where, DMEM (Dulbecco's Modified Eagle Medium) is the most suitable medium for the culture of cells and tissues for many adherent cell phenotypes among defined media. The modification of Dulbecco is an enhanced additional formulation that boosts up to four-fold the selection of amino acid and vitamin content of the original medium of Eagle.

#### **3.4.2.2 10 % Fetal Bovine Serum**

50 mL of Fetal Bovine Serum was added to the 500 mL of DMEM to prepare 10 % FBS. FBS is high in albumin that provides nutrient and adequate environment to grow the cells more quickly.

#### **3.4.2.3 2% DMSO Solution**

2 % DMSO solution was prepared by adding  $600 \mu l$  in  $29.4 \ mL$  of distilled water for negative control.

#### **3.4.2.4 Trypsin**

0.25 % trypsin was used in the medium.

#### 3.4.2.5 Used Consumables

96 well plate, 15 mL tubes, Tips, Gloves, PTFE Syringe Filter, Culture flask, Cell culture media, 1% Penicillin-Streptomycin, Gentamycin, Serological pipette, Trypsin etc.

#### 3.4.2.6 Used Instruments

Biological Bio Safety Cabinet (Model: NU-400E, Nuaire, USA), CO2 incubator (Nuaire, USA), trinocular Microscope with Camera (Olympus, Japan), hemocytometer.

#### 3.4.2.7 Celltiter 96 Assay Kit

Celltiter 96 assay kit is an accumulation of qualified reagents that is a fast and advantageous system for the determination of the amount of proliferation and cytotoxicity (Ifere et al., 2010). Under a formazan product, it shows the view of change of a cell division with tetrazolium salt, that is undoubtedly distinguished utilizing a 96 well plate. A mixer of dye solution was added in 96 well plates to culture cells to perform this assay.

# 3.4.3 Preparation of the Different Concentrations of Plant Extract

The assay was conducted by using 5 concentration 25 mg/ mL, 2.5 mg/ mL, 0.25 mg/ mL and 0.025 mg/ mL and 0.0025 mg/ mL of Triphala extract in the ratio 1:1:1 (A:B:H) and 2:1:1 (A:B:H), where Amlaki was used in twice the amount. 25 mg/ mL concentration was made by placing in 25 mg of extract of each (total 75 mg) in 3 mL DMSO and it was the store solution. 2.5 mg/ mL concentration was completed by diluting

25 mg/ mL solution 10 times by DMSO. 30  $\mu$ L of sample 1 is put in to the 270  $\mu$ L of DMSO to make 2.5 mg/ mL concentration. 0.25 mg/ mL and 0.025 mg/ mL concentrations were completed by sequential dilution with. Later, the samples were sieved using a syringe filter of 0.45  $\mu$ m pore size.

#### 3.4.4 Cell Culture

#### 3.4.4.1 Preparation of Assay Plates

N4X4 cell lines were maintained in RPMI and DMEM (Dulbecco's Modified Eagle's Medium) in addition with 1 % penicillin-streptomycin respectively, 0.2 % Gentamicin, 10 % fetal bovine serum.

#### 3.4.4.2 Thawing of Cells

N4X4 cells were preserved in liquid nitrogen in cryovials. The cryovial was taken and rapidly defrosted by swirling the vial delicately using a water bath at 37°C until there was a small piece of ice left in the vial. Afterward the thawed cells were transferred into a centrifuge tube which contained the DMEM medium drop wise under a laminar airflow hood. Thereafter, the cells were suspended in the medium gently and transferred to the culture vessels. 90  $\mu$ L of DMSO to make 2.5 mg/ mL concentration. 0.25 mg/ mL and 0.025 mg/ mL concentrations were completed by sequential dilution with. Later, the samples were sieved.

# 3.4.4.3 Cell Passage

To get a fresh cell suspension cell passaging was done by transferring the cells into a new medium. Thus, cultured media was washed by FBS and followed by addition of  $800 \, \mu l$  of trypsin for detaching the cells from the top of the culture vessels. Then the cells were incubated and checked for the detachment under a microscope. After watching  $90 \, \%$  of cells detached,  $5 \, mL$  DMEM media was added to the vessels and blended using

a pipette. Finally, 1 mL of this solution was taken and mixed with 4 mL of DMEM in a new vessel and kept in an incubator.

#### 3.4.4.4 Harvesting of Cells

The cells were harvested using trypin in log phase growth.

# 3.4.4.5 Counting of Cells

Hemocytometer was used for cell counting. The hemocytometer was prepared by cleaning and polishing the mirror like surface deliberately with ethanol and lens paper. The coverslip was put in the counting surface and then, using a pasteur pipet, the fresh cell suspension was introduced into the hemocytometer. Enough suspension was introduced for the surface to be merely overflowed. Afterward the counting chamber was set in the microscope stage then the counting grid was focused. In a standard hemocytometer (figure 7) with rulings 1 entire grid can be observed at 40X magnification. The cells present in the 4 large squares were counted afterwards. Either upper and left sides touching cells or lower or right sides touching cells were counted.

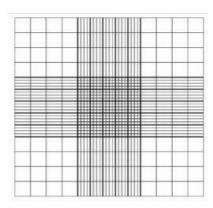


Figure 7: Hemocytometer

#### 3.4.5 Procedure

Cytotoxic effect was performed in the Centre for Advanced Research in the Sciences (CARS) using their services. The celltiter 96 non- radioactive cell proliferation assay kit was used to perform the MTT colorimetric assay (Promega, USA). Cells were seeded onto 96 well plates and incubated at 37 °C and 5% carbon dioxide atmosphere. After 24 hours of incubation, 10µL of sample was added into each well. Then, 2 days of again incubation took place. After 2 days of incubation, cytotoxicity was examined using the assay kit and the absorbance was measured at 570 nm using a 96 Well Plate Reader. Same process was followed for the negative control and the 5 concentrations. Negative control contained medium with 2% DMSO solution and blank contained only the medium. Duplicate wells were used for each sample and finally the cytotoxic activity was calculated by using a formula which is given below:

% of cytotoxic activity

$$= 100 - \frac{Absorbance of test sample}{Absorbance of negative control} \times 100$$

# Chapter 4

# **Results**

# 4.1 Determination of Percentage Yield of the Plant Extract Triphala

# **4.1.1 Total Weight of Powdered Plant Material**

Table 8 represents the initial weight of powdered fruits used, the final extract and the % yield.

Table 8: % yield of Amlaki, Bibhttaki and Haritaki extracts

Sample	Initial weight of powder	Final weight of Extract	% Yield
Amlaki	600 g	88.23 g	14.71 %
Bibhitaki	600 g	55.33 g	9.22 %
Harikati	600 g	97.97 g	16.33 %

**Interpretation:** The % yield for Haritaki was the most compared to Amlaki and Bibhitaki. This is because more compound from Haritaki dissolved in the solvent used, which is methanol.

# 4.1.2 Calculation of Percentage Yield of the Extract

% yield = 
$$\frac{W1}{W2} \times 100$$

Where,  $W_1$  = net weight of extract after maceration in gram

 $W_2$  = total weight of powder taken for extraction in gram

# Percentage Yield of Bibhitaki

Percentage yield =

$$\frac{55.328}{600} \times 100$$

# Percentage Yield of Amlaki

Percentage yield =

$$\frac{88.23}{600} \times 100$$

# Percentage Yield of Haritaki

Percentage yield =

$$\frac{97.965}{600} \times 100$$

**Interpretation:** The percentage yield of leaf extract was found to be 9.22%, 14.71% and 16.33% from 600g of Bibhitaki, Amlaki and Haritaki respectively after the maceration process.

# 4.2 DPPH (1, 1-diphenyl-2-picryl hydroxyl) Free Radical Scavenging Assay

# 4.2.1 Table of DPPH free Radical Scavenging Activity of Ascorbic Acid, Bibhitaki, Amlaki, Haritaki and Different Ratios of Triphala

The concentration and absorbance values of sample and standard (Ascorbic Acid) are given below in table 9:

Table 9: Table of DPPH Free Radical Scavenging Assay at 517 nm (Absorbance vs. concentration) of ascorbic acid, Bibhitaki, Amlaki and Haritaki

Concentration (µg/ mL)	Absorbance of Ascorbic Acid (Standard)	Absorbance of Bibhitaki	Absorbance of Amlaki	Absorbance of Haritaki
3.125	0.608	0.647	0.598	0.611
6.25	0.457	0.447	0.438	0.583
12.5	0.312	0.275	0.256	0.283
25	0.129	0.134	0.088	0.178
50	0.053	0.052	0.043	0.048
100	0.039	0.042	0.029	0.039
200	0.029	0.033	0.028	0.038
400	0.029	0.031	0.026	0.033
800	0.024	0.031	0.025	0.031
1200	0.023	0.032	0.024	0.030

The Table 9 above shows the absorbance values of Bibhitaki, Amlaki, Haritaki and standard corresponding to their respective concentrations at 517nm. The standard deviations are also stated. The concentrations are ranging from  $3.125\mu g/$  mL to  $1200\mu g/$  mL.

Table 10: Table of DPPH Free Radical Scavenging Assay at 517 nm (Absorbance vs. concentration) of the different ratios of Triphala

Concentration (μg/ mL)	Absorbance of 1:1:1 (A:B:H)	Absorbance of 2:1:1 (A:B:H)	Absorbance of 1:2:1 (A:B:H)	Absorbance of 1:1:2 (A:B:H)
3.125	0.564	0.548	0.626	0.694
6.25	0.426	0.370	0.396	0.354
12.5	0.326	0.262	0.173	0.227
25	0.127	0.104	0.109	0.147
50	0.058	0.053	0.030	0.037
100	0.026	0.018	0.027	0.035

The Table 10 above shows the absorbance values of the different Triphala ratios corresponding to their respective concentrations at 517nm. The standard deviations are also stated. The concentrations are ranging from 3.125  $\mu$ g/ mL to  $100\mu$ g/ mL.

# 4.2.2 Graph of Absorbance vs. Concentration of Ascorbic Acid, Bibhitaki, Amlaki, Haritaki and Different Ratios of Triphala in DPPH free radical scavenging assay

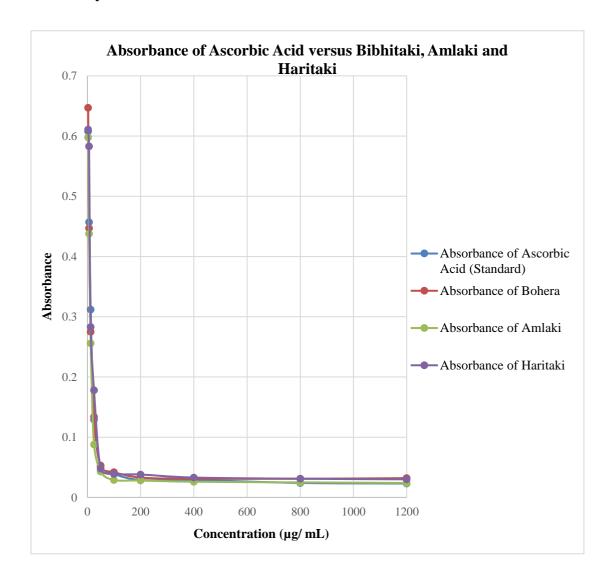


Figure 8: Graph of Absorbance vs. Concentration (µg/mL) for Ascorbic Acid, Bibhitaki, Amlaki and Haritaki in DPPH free radical scavenging assay

**Interpretation:** From the Table 9 and Figure 8 above, it was observed that with an increase in concentration of extracts from  $3.125-1200\mu g/mL$ , the absorbance decreased gradually. However, after  $100\mu g/mL$  the absorbance decrease was not as drastic as before and only small differences could be seen. Moreover, the graphs of all the extracts have similar slope for Absorbance vs. Concentration ( $\mu g/mL$ ).

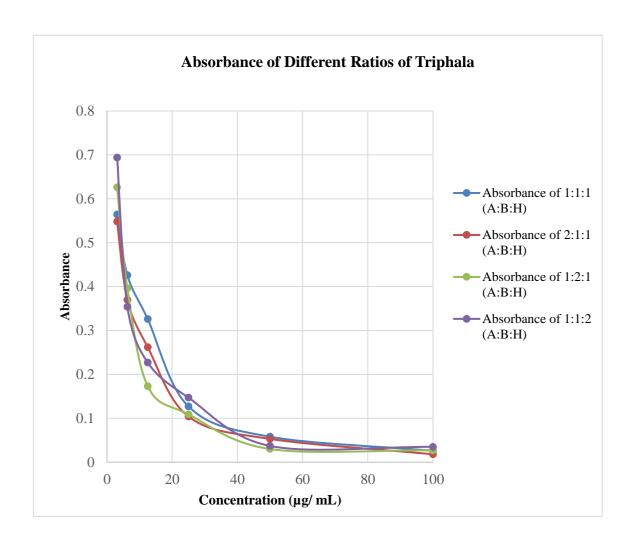


Figure 9: Graph of Absorbance vs. Concentration (µg/mL) for different ratios of Triphala in DPPH free radical scavenging assay

**Interpretation:** In Figure 9 it was observed that with an increase in concentration from  $3.125-100\mu g/mL$ , the absorbance decreased gradually for all the ratios. The absorbance of 1:2:1 (A:B:H) has the steepest slope shown in green compared to the standard shown in blue.

# 4.2.3 Table of Percentage of Inhibition of Ascorbic Acid, Bibhitaki, Amlaki, Haritaki and different ratios of Triphala

Table 11: Table of % Inhibition of DPPH Free Radical by Ascorbic Acid, Bibhitaki, Amlaki and Haritaki

Concentration (μg/ mL)	% inhibition of Ascorbic Acid (Standard)	% inhibition of Bibhitaki	% inhibition of Amlaki	% inhibition of Haritaki
3.125	20.209	15.092	21.522	19.816
6.25	40.0263	41.339	42.519	23.491
12.5	59.055	63.911	66.404	62.861
25	83.071	82.415	88.451	76.640
50	93.045	93.176	94.357	93.701
100	94.882	94.488	96.194	94.882
200	96.194	95.669	96.325	95.013
400	96.194	95.722	96.588	95.669
800	96.850	95.831	96.719	95.932
1200	96.982	95.841	96.850	96.063

**Interpretation**: In Table 11 it was observed that for concentrations 3.125 to 100  $\mu$ g/mL, the % inhibition of free radical scavengers gradually increased. However, from 200 to 1200  $\mu$ g/ mL, the values for both the standard and samples remained nearly constant. 95.801 %, 96.850% and 96.063% of inhibition of free radical was found for Bibhitaki, Amlaki and Haritaki respectively, at 1200  $\mu$ g/ mL.

Table 12: Table of % Inhibition of DPPH Free Radical by different ratios of Triphala

Concentration (µg/ mL)	% inhibition of 1:1:1 (A:B:H)	% inhibition of 2:1:1 (A:B:H)	% inhibition of 1:2:1 (A:B:H)	% inhibition of 1:1:2 (A:B:H)
3.125	25.984	28.083	17.847	8.923
6.25	44.094	51.443	48.031	53.543
12.5	57.218	65.616	77.296	70.209
25	83.333	86.351	85.695	80.708
50	92.388	93.044	96.062	95.144
100	96.587	97.637	96.456	95.406

Interpretation: In Table 12 it was observed that for concentrations  $3.125 \,\mu\text{g/}$  mL to  $100 \,\mu\text{g/}$  mL the % inhibition of free radical scavengers by ascorbic acid and the extract ratios gradually increased. The highest free radical scavenging activity for the sample ratios were found to be at  $100 \mu\text{g/}$  mL and the percentage inhibition was 96.587%, 97.637%, 96.456% and 95.406% for the Triphala ratios 1:1:1(A:B:H), 2:1:1 (A:B:H), 1:2:1 (A:B:H) and 1:1:2 (A:B:H) respectively.

# 4.2.4 Graph of % Inhibition of DPPH Free Radical by Ascorbic Acid, Bibhitaki, Amlaki, Haritaki and Different Ratios of Triphala

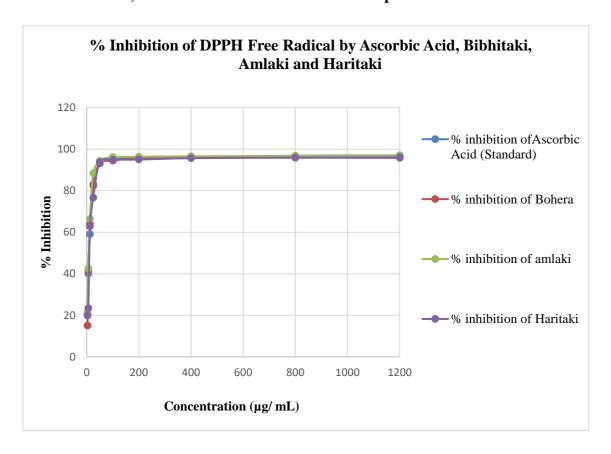


Figure 10: Graph of % Inhibition of DPPH Free Radical by Ascorbic Acid, Bibhitaki, Amlaki and Haritaki

**Interpretation**: In Figure 10 we can see the % inhibitory activity has increased from concentration ranging from 3.125 to  $100\mu g/$  mL. whereas the increase was not significant from concentration 100 to  $1200\mu g/$  mL. In conclusions, it can be said that with increasing concentrations, the percentage of inhibition did not increase proportionally. All the extracts have similar slope of % inhibition.

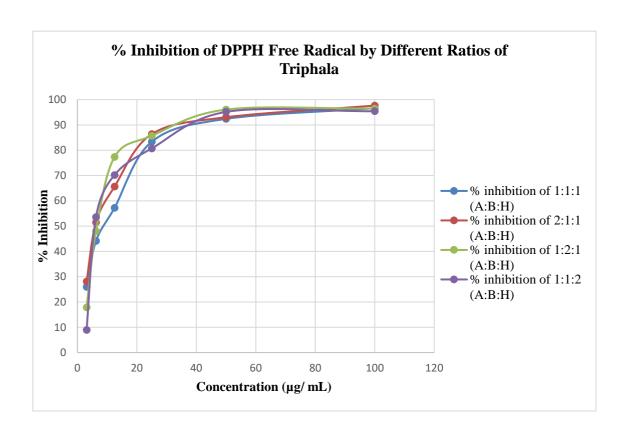


Figure 11: Graph of % Inhibition of DPPH Free Radical by different ratios of Triphala

**Interpretation**: In Figure 11 we can see the % inhibitory activity has increased from concentration ranging from 3.125 to  $100\mu g/$  mL. In conclusions, it can be said that with increasing concentrations, the percentage of inhibition did increase proportionally. The % inhibition of 1:2:1 (A:B:H) shown in green has the steepest slope until it starts to decline.

# 4.2.5 Determination of the $IC_{50}$ of Ascorbic Acid, Bibhitaki, Amlaki, Haritaki and Different Ratios of Triphala

Table 13: IC<sub>50</sub> values of Ascorbic Acid, Bibhitaki, Amlaki, Haritaki and Different Ratios of Triphala

Samples and Standard	IC <sub>50</sub> value in μg/ mL
Ascorbic acid (Standard)	12.96
Haritaki	15.93
Bibhitaki	12.99
Amlaki	10.287
1:1:1 (A:B:H) ratio	11.28
2:1:1 (A:B:H) ratio	7.01
1:2:1 (A:B:H) ratio	8.33
1:1:2 (A:B:H) ratio	11.03

Interpretation: In Table 13 it can be observed that 15.93  $\mu$ g/ mL, 12.99  $\mu$ g/ mL and 10.287  $\mu$ g/ mL of Haritaki, Bibhitaki and Amlaki was found to inhibit 50 % of DPPH free radical scavengers, which is the IC<sub>50</sub> value. As the IC<sub>50</sub> value for Ascorbic Acid was 12.96  $\mu$ g/ mL, Amlaki showed more inhibition at a lower concentration. The IC<sub>50</sub> value for the Triphala ratios 1:1:1 (A:B:H), 2:1:1 (A:B:H), 1:2:1 (A:B:H) and 1:1:2 (A:B:H) were 11.28  $\mu$ g/ mL, 7.01  $\mu$ g/ mL, 8.33  $\mu$ g/ mL and 11.03  $\mu$ g/ mL respectively and it was observed that all the values were lower than that of the standard and the individual components of Triphala, with 2:1:1 (A:B:H) being the lowest and most inhibitory. This suggested synergistic effect when the three extracts are mixed.

# 4.3 Total Phenolic Content (TPC) Determination

# 4.3.1 Calibration Curve of Gallic Acid at 765 nm

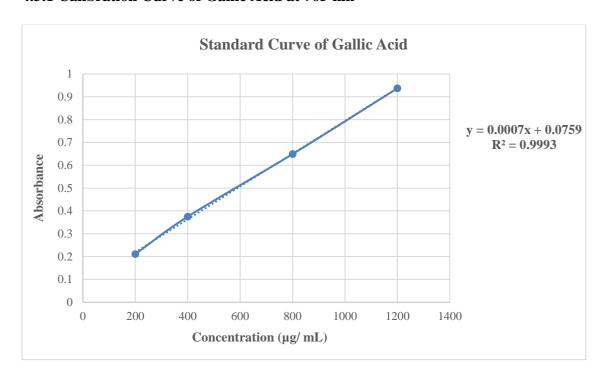


Figure 12: Standard curve of standard gallic acid at 765 nm to determine the TPC in sample

# 4.3.2 Table of total phenolic content in Triphala

Table 13: Total phenolic content (TPC) of Bibhitaki is marked as gallic acid equivalent (GAE)

Concentration of	Total Phenolic	Total Phenolic	Total Phenolic
	Content at	Content at	Content at
Sample (μg/ mL)	765 nm of Bibhitaki	765 nm of Amlaki	765 nm of Haritaki
200	190.66	178.9	37.51
400	210.37	243.94	220.66
800	349.66	348.23	254.66
1200	368.94	395.09	260.65

Interpretation: It is observed from Table 13 and Figure 12 that, concentration of samples has proportional relationship with the TPC. When the concentration (200  $\mu$ g/ mL to 1200  $\mu$ g/ mL) of methanol extract of samples increased, total phenolic content also increased from 190.66 mg to 368.94 mg, 178 mg to 395.09 mg and 37.51 mg to 260.65 mg of gallic acid equivalent per gram of dried extract for Bibhitaki, Amlaki and Haritaki respectively. Moreover, it indicates that with the increase of total phenolic content, its antioxidant activity also increases.

# **4.4**Total Antioxidant Capacity Determination

#### 4.4.1 Calibration curve of Ascorbic Acid (Standard) at 695 nm

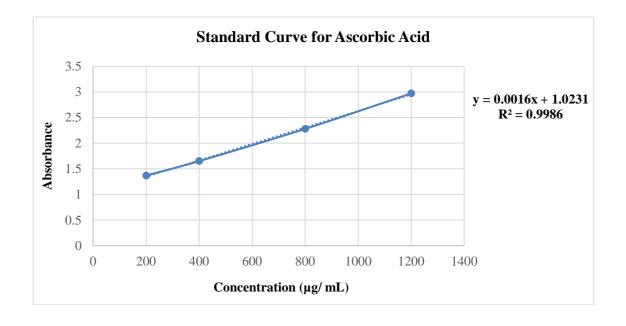


Figure 13: Standard curve of ascorbic acid at 695 nm for determining TAC value in methanol extract of Triphala

# 4.4.2 Table of Total Antioxidant Capacity (TAC) in Triphala

Table 14: Total antioxidant content (TAC) of Triphala is marked as Ascorbic Acid equivalent (AAE)

Concentration	Total Antioxidant	Total Antioxidant	Total Antioxidant
of Sample	Capacity (TAC) at	Capacity (TAC) at	Capacity (TAC) at
(μg/ mL)	695 nm of Bibhitaki	695 nm of Amlaki	695 nm of Haritaki
200	8.51	16.95	90.075
400	121.39	163.2	154.76
800	197.14	235.95	417
1200	245.32	485.5	559.76

Interpretation: It is observed from the table 14 and figure 13, concentration of *Triphala* has proportional relationship with the TAC. When the concentration (200  $\mu$ g/ mL to 1200  $\mu$ g/ mL) of methanol crude extract of Triphala increased, total antioxidant capacity also increased from 8.51 to 245.32 mg, 16.95 to 485.5 mg and 90.075 to 559.76 mg of ascorbic acid equivalent per gram of dried extract for Bibhitaki, Amlaki and Haritaki respectively. Moreover, it indicates that with the increase of total flavonoid content, its antioxidant activity also increases.

# 4.5 In vitro Cytotoxicity Test of Triphala by MTT Assay on N4X4 (Human Glioma) Cell Line

The cytotoxicity activity of methanol extract of Triphala in 1:1:1 (A:B:H) and 2:1:1 (A:B:H) ratio was accomplished by MTT assay on N4X4 cell line. Different concentrations (25 mg/ mL, 2.5 mg/ mL, 0.25 mg/ mL and 0.025 mg/ mL) of the methanol extract of sample were used to investigate the cytotoxic activity. 2 % DMSO in DMEM medium was used as a negative control. Absorbance along with their average

was observed for each of the concentrations. Survival of cells and % of growth inhibition of N4X4 cell line in different concentrations along with their IC<sub>50</sub> value are given in the table 15. Images were also captured which are attached in figure 14.

# 4.5.1 Cytotoxicity Test on N4X4 Cell Line using 1:1:1 (A:B:H) ratio

Table 15: Survival and inhibition of N4X4 cell line in different concentrations along with their IC50 value and average absorbance using the 1:1:1 (A:B:H) ratio

Sample Concentration (mg/ mL) 1:1:1 (A:B:H) ratio	Average Absorbance	Survival of N4X4 Cell (%)	N4X4 Cell Growth Inhibitions (%)	IC <sub>50</sub> Value (mg/ mL)
2% DMSO (cell + solvent)	3.754	100	0	
25	0.418	11.12	88.88	
2.5	3.308	88.11	11.89	13.43
0.25	3.551	94.57	5.43	
0.025	3.589	95.61	4.39	
0.0025	3.606	96.04	3.96	

**Interpretation:** Table 15 shows the percentage of the survival of the N4X4 cell that was observed through the trinocular microscope. When the concentration of sample extract was 25 mg/ mL, highest cell death was detected (88.88 % cell death where survival of N4X4 cells were 11.12 %) after 48 hours of incubation. Extracts exhibited weak cytotoxic activity at the other four concentrations of 2.5 mg/ mL, 0.25 mg/ mL, 0.025 mg/ mL and 0.0025 mg/ mL, as the percentage of inhibition was 11.89%, 5.43%, 4.39% and 3.96 % respectively.

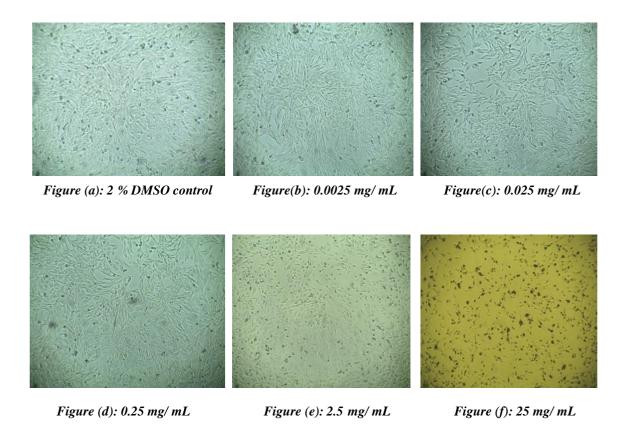


Figure 14: Cell viability of methanol extract of Triphala, 1:1:1(A:B:H) ratio, at different extract concentrations of 0.0025 mg/mL, 0.025mg/mL, 0.25 mg/mL, 2.5 mg/mL, 25 mg/mL and 2% DMSO as control respectively after incubating 48 hours in N4X4 cell line

Interpretation: Figure 14 illustrates the cell viability of N4X4 cells in different concentrations of the plant extracts as well as in the control (2 % DMSO). Figure (a) represents the control where 2 % DMSO was added and no plant extract was included. In the figure (b), cell viability decreased due to adding leaf extract of 0.0025 mg/ mL concentration; only 3.96 % cell growth inhibition was perceived. Slightly less cell viability was observed while 0.025 mg/ mL and .25 mg/ mL concentration of the extract was added in figure (c) and (d). In the next picture figure (e), visible cell growth inhibition was observed due to adding 2.5 mg/ mL concentration where percentage of cell growth inhibition was 11.89%. Finally, in the last picture, highest cytotoxicity of plant extract was observed as it killed maximum number of cells and percentage of cell

growth inhibition was highest; approximately 88.88 %. It depicts that, plant extract of 25 mg/ mL concentration shows a remarkable cytotoxic potential with 88.88 % cell inhibition.

#### 4.5.2 Cytotoxicity Test on N4X4 Cell Line using 2:1:1 ratio (A:B:H)

Table 16: Survival and inhibition of N4X4 cell line in different concentrations along with their IC<sub>50</sub> value and average absorbance using 2:1:1 (A:B:H) ratio

Sample Concentration (mg/ mL) 2:1:1 (A:B:H) ratio	Average Absorbance	Survival of N4X4 Cell (%)	N4X4 Cell Growth Inhibition (%)	IC50 (mg/ mL)
2% DMSO (cell + solvent)	3.754	100	0.00	
25	0.439	11.25	88.75	
2.5	0.364	85.48	14.52	13.23
0.25	3.209	95.07	4.93	
0.025	3.644	97.06	2.94	
0.0025	3.753	99.98	0.02	

**Interpretation:** Table 16 shows the percentage of the survival of the N4X4 cell that was observed through the trinocular microscope. When the concentration of sample extract was higher (25 mg/ mL), highest cell death was detected of 88.75 %, where survival of N4X4 cells were 11.25 % after 48 hours of incubation. Extracts exhibited weak cytotoxic activity at the other four concentrations of 2.5 mg/ mL, 0.25 mg/ mL, 0.025 mg/ mL and 0.0025 mg/ mL, as the percentage of inhibition was 14.52%, 4.93%, 2.94% and 0.02 % respectively.

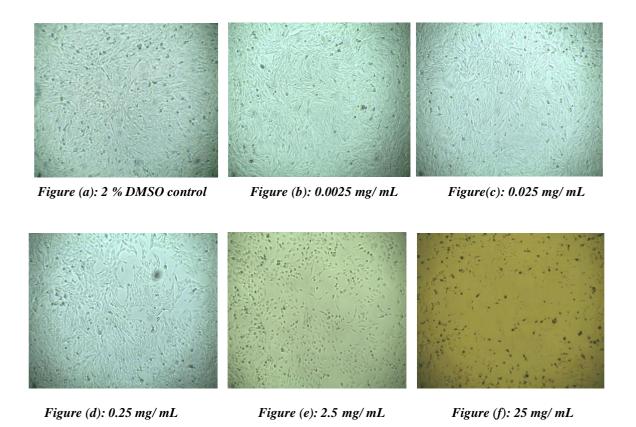


Figure 15: Cell viability of methanol extract of Triphala, 2:1:1 (A:B:H) ratio, at different extract concentrations of 0.0025 mg/mL, 0.025mg/mL, 0.25 mg/mL, 2.5 mg/mL, 25 mg/mL and 2% DMSO as control respectively after incubating 48 hours in N4X4 cell line

Interpretation: Figure 15 illustrates the cell viability of N4X4 cells in different concentrations of the plant extracts as well as in the control (2 % DMSO). Figure (a) represents the control where 2 % DMSO was added and no plant extract was included. In the figure (b), cell viability decreased due to adding leaf extract of 0.0025 mg/ mL concentration; only 0.02 % cell growth inhibition was perceived. Slightly less cell viability was observed while 0.025 mg/ mL and .25mg/ mL concentration of the extract was added in figure (c) and (d). In the next picture figure (e), visible cell growth inhibition was observed due to adding 2.5mg/ mL concentration where percentage of cell growth inhibition was 14.52%. Finally, in the last picture, highest cytotoxicity of plant extract was observed as it killed maximum number of cells and percentage of cell

growth inhibition was highest; approximately 88.75 %. It depicts that, plant extract of 25 mg/ mL concentration shows a remarkable cytotoxic potential with 88.75 % cell inhibition.

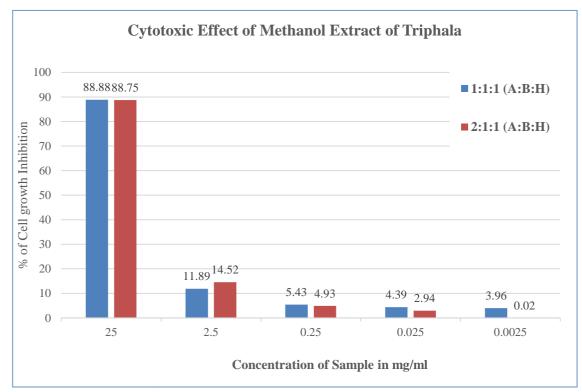


Figure 16: Cytotoxic effect of methanol crude extract on N4X4 cell line

**Interpretation:** The figure 16 complies the graphical representation of the percentage of cell growth inhibition vs. the respective concentration samples. From the bar chart, it was observed that the lowest percentage of cell inhibition was at 0.0025 mg/ mL concentration of plant extract. On the other hand, the highest percentage of cell inhibition was observed at 25 mg/ mL concentration of plant extract. The % inhibition of the two ratios of Triphala are almost the same and so change in concentration of Amlaki did not increase % inhibition of cell growth.

# Chapter 5

# **Discussion**

This study was conducted with Triphala, which is a mixture of Amlaki, Bibhitaki and Haritaki. 600 gm of individual fruit powder of Amlaki, Bibhitaki and Haritaki were soaked in methanol and the weight of extract after evaporation and drying was 88.23g, 55.33g and 97.97g, thus giving a percentage yield of 14.71%, 9.22% and 16.33% respectively. The higher percentage yield of Haritaki is because more compounds were able to dissolve in methanol. Moreover, methanol solvent was never used before to estimate the antioxidant activity by TAC and TPC and DPPH assay was not conducted on different ratios of Triphala to compare with the equivalent mixture. That is why it is one of the appropriate choices to determine the antioxidant potential of methanol extract. In vitro antioxidant tests such as DPPH free radical scavenging assay, Total Phenolic Content (TPC) and Total Antioxidant Capacity (TAC) were carried out to examine the antioxidant potential of methanol extract of Amlaki, Bibhitaki and Haritaki. Furthermore DPPH free radical scavenging assay was also carried out for different ratios of Triphala which were, 1:1:1 (A:B:H), 2:1:1 (A:B:H), 1:2:1 (A:B:H) and 1:1:2 (A:B:H). In DPPH test, the highest percentage of inhibition of free radical was found at the highest concentration of 1200 μg/ mL which was 95.801 %, 96.850% and 96.063% of inhibition of free radical was found for Bibhitaki, Amlaki and Haritaki. Furthermore, at 100 µg/ mL, which is the highest concentration, the percentage inhibition was 96.587%, 97.637%, 96.456% and 95.406% for the Triphala ratios 1:1:1 (A:B:H), 2:1:1 (A:B:H), 1:2:1 (A:B:H) and 1:1:2 (A:B:H) respectively. As the percentage inhibition of the ascorbic acid (standard) was 96.982%, only 2:1:1 (A:B:H) Triphala ratio showed more inhibition which was 97.637%. From the perspective of IC<sub>50</sub> value, 15.93 µg/ mL, 12.99 µg/ mL and 10.287 µg/ mL of Haritaki, Bibhitaki and Amlaki was found to inhibit 50 % of DPPH free radical scavengers. As the IC<sub>50</sub> value for Ascorbic Acid was 12.96 µg/ mL, Amlaki showed more inhibition at a lower concentration. The IC<sub>50</sub> value for the Triphala ratios 1:1:1(A:B:H), 2:1:1 (A:B:H), 1:2:1 (A:B:H) and 1:1:2 (A:B:H) were 11.28 µg/ mL, 7.01 µg/ mL, 8.33 µg/ mL and 11.03 µg/ mL respectively and it was observed that all the values were lower than ascorbic acid and the individual components of Triphala as well. The ratio 2:1:1 (A:B:H) gave the lowest value and was the most inhibitory. This suggested synergistic effect when the three extracts are mixed together.

In total phenolic test (TPC) and total antioxidant test, gallic acid and ascorbic acid was used as the standard. In case of total phenolic test, at concentration of  $1200~\mu g/$  mL, dried extract showed highest amount which was 368.94mg, 395.09mg and 260mg of gallic acid per gram for Bibhitaki, Amlaki and Haritaki respectively. Finally, for total antioxidant test, at the concentration of  $1200~\mu g/$  mL, dried extract showed highest amount which was 245.32mg, 485.5mg and 559.76mg of ascorbic acid per gram for Bibhitaki, Amlaki and Haritaki respectively. From the values obtained from the tests, we can conclude that Amlaki had the highest gallic acid per gram of 395.09mg and Haritaki had the highest ascorbic acid per gram of 559.76mg.

Additionally, cancer cell lines were selected for screening the cytotoxic activity before such as Michigan Cancer Foundation-7, breast cancer cells, prostate cancer cells, human pancreatic cancer cells, mouse breast cancer cells and mouse thymic lymphoma (Kaur et al., 2005; Sandhya et al., 2006; Sandhya & Misha, 2006; Kaur et al., 2005, Shi et al., 20008; Sandhya et al., 2006). Among all of the solvents used for screening, methanol extract showed highest percentage in case of cytotoxicity potentiality. However, N4X4 cell line related to brain cancer were not used previously in order to

check the cytotoxic activity of Triphala. Furthermore, the ratio 2:1:1 (A:B:H) was also used alongside the ratio 1:1:1 (A:B:H) to compare and see if using more Amlaki, which has the highest gallic acid among the three and is also more easily edible, would show more cytotoxicity. Gallic acid is one of the major components of Triphala and capable of inhibiting cancer cell proliferation suggesting the key factor responsible for antimutagenic and cytotoxic effects of Triphala (Kaur et al., 2005).

Afterwards, *in vitro* cytotoxicity test known as MTT assay was conducted on N4X4 cell line where cytotoxicity was examined after 48 hour of incubation and 2% DMSO was used as the negative control. C and percentage of cell growth inhibition was determined. and 25 mg/ mL concentration of extract provided the highest cytotoxic activity where percentage of cell death was 88.88 % for 1:1:1 (A:B:H) and 88.75% for 2:1:1 (A:B:H). Data interpretation also revealed that 50 % of cell death required 13.43 mg/ mL in case of 1:1:1 (A:B:H) and 13.23 mg/ mL for 2:1:1 (A:B:H) of extract concentration meaning the extract showed moderate cytotoxic activity in N4X4 cell line. The percentage of cell growth inhibition was similae for both the ratios and so, more Amlaki did not give better results as their cytotoxic activity was almost the same with a small difference of .20 mg.

Triphala (MET) exhibited dose dependent antiproliferative properties in cancer stem cells (HCCSC) and cancer cells (HCT116) and apoptotic properties in HCCSCs. Apoptosis induction by Triphala was via the mitochondrial apoptotic signaling characterized by elevated Bax/Bcl-2 ratio. These results indicate that Triphala (MET) may serve as an effective anticancer agent, mainly against colon cancer and has a potential to be taken simultaneously to conventional chemotherapy in the management of colon cancer (Vadde et al., 2015). The discrepancies however may be due to usage of different *in-vitro* methods, variation in concentration and other physiological factors

which lead to less activity against N4X4 cancer cell line. On a concluding note, methanol extract of Triphala confirms highest antioxidant activity and moderate cytotoxic activity. So, methanol extract can be used as a promising antioxidant agent, but it is not highly suggested as anticancer agent to treat brain tumour.

## Chapter 6

## **Conclusion**

In DPPH test, highest percentage of inhibition of free radical was found at the highest concentration which was 1200  $\mu$ g/ mL. The IC<sub>50</sub> values were 15.93  $\mu$ g/ mL, 12.99  $\mu$ g/ mL and 10.287  $\mu g/$  mL of Haritaki, Bibhitaki and Amlaki, for Ascorbic Acid it was 12.96  $\mu$ g/ mL. The IC<sub>50</sub> value for the Triphala ratios 1:1:1 (A:B:H), 2:1:1 (A:B:H), 1:2:1 (A:B:H) and 1:1:2 (A:B:H) were 11.28  $\mu$ g/ mL, 7.01  $\mu$ g/ mL, 8.33  $\mu$ g/ mL and 11.03 µg/ mL respectively and it was observed that all the values were lower than that of the standard and the individual components of Triphala. The 2:1:1 (A:B:H) showed the lowest value among all the samples. This suggested synergistic effect when the three extracts are mixed together. Thus, this provides the evidence that small amount of concentrations of is required to inhibit the DPPH free radicals and shows strong antioxidant activity against free radicals. Total phenolic content and total antioxidant capacity were also found strong compared to the standard through the conducted experiments. Thus, Triphala showed high antioxidant activity in all of the testes. Additionally, N4X4 cell line was used to investigate the cytotoxic potential of Triphala and moderate cell toxicity was observed at 25 mg/ mL concentration of the two ratios. To conclude, methanol extract of Triphala signifies high antioxidant activity and moderate cytotoxic potential and can be used to treat oxidative degenerative diseases but cannot be used effectively for brain cancer until further refinement of useful

compounds from this extract can be done.

## **Future Direction**

The result from this study established that methanol extract of Triphala implied very strong antioxidant activity compared to the standard ascorbic acid. On the other hand, Triphala possessed moderate cytotoxic activity.

- Other ratios of the Triphala extract can be used to find cytotoxicity
- Triphala extract can be used in different ratios to see it's chemoprotective and radioprotective capacity
- Synergistic outcomes can be highlighted properly with the use of methanol extract of Triphala.
- Other plant parts such as root, stem can be used to investigate the cytotoxic and antioxidant potential using the same methanol solvent.
- High amount of antioxidant activity shows scope to further isolate and analyze new compounds.

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